



Plasmon assisted isomerisation in naphthalene and azulene

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Motivation



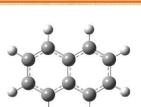
Two extremely bright stars illuminate a mist of PAHs in this Spitzer image (NASA/JPL-Caltech/ZMASS/SSI/University of Wisconsin)



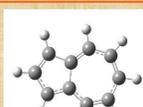
Terrestrial sources of PAHs: industries and automobiles
http://blog.chemistry-matters.com

- Astrochemistry of ISM is a function of dissociative and isomerization processes in PAHs and their derivatives.
- Molecular dynamics of PAHs is a benchmark for several biologically important processes.
- 6 to 5 membered ring deformation is an important pathway of fullerene formation in ISM.
- Neutral elimination of PAH cations strongly depends on isomerisation.
- Azulene and Naphthalene are the simplest isomeric pairs with special attributes.
- Plasmon resonance as a dominant and constant energy loss mechanism in ion-PAH collision.

Isomerisation

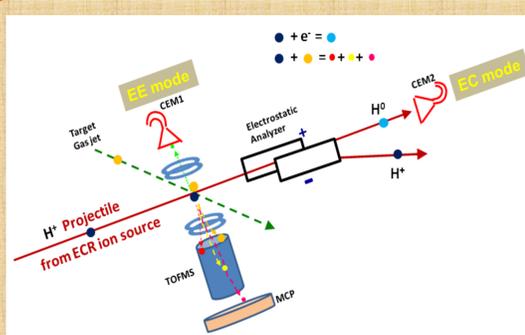


Naphthalene



Azulene

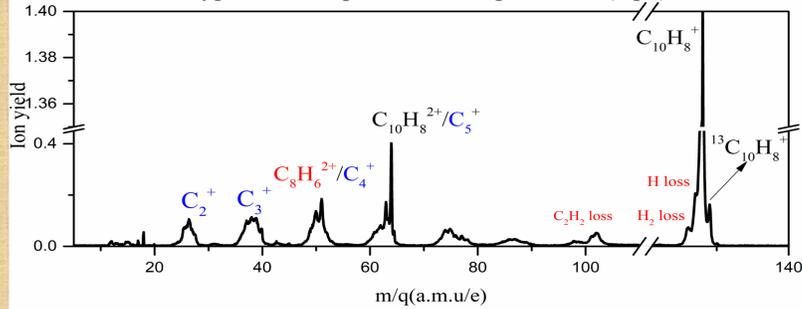
Experimental setup



- Low Energy Ion Beam Facility, IUAC, New Delhi
- Energy of proton beam: 50 keV to 150keV (intermediate velocity).

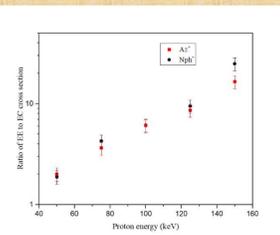
- Target molecules:
 - Naphthalene (C₁₀H₈),
 - Azulene (C₁₀H₈),
 - Quinoline (C₁₀H₇N),
 - Iso-quinoline (C₁₀H₇N)
 - Fluorene (C₁₃H₁₀)
- Recoil ions were detected by a position sensitive MCP detector after extraction by a linear two field ToF mass spectrometer.
- Multi-hit coincidence measurement were done in both electron emission & electron capture mode.

Typical mass spectrum of Naphthalene (Nph)



- Sharp peaks show intact single or multiply charged parent ion and neutral evaporation peaks whereas small fragment mass peaks are poorly resolved.
- Multiple H loss is observed with decreasing order of yield.
- Dication yield is accompanied by mostly 2H/H₂ loss.
- Dication is followed by strong C₂H₂ loss with again mostly 2H/H₂ loss.
- Azulene shows almost identical spectrum but with relatively larger fraction of C₂H₂ loss from mono cation.

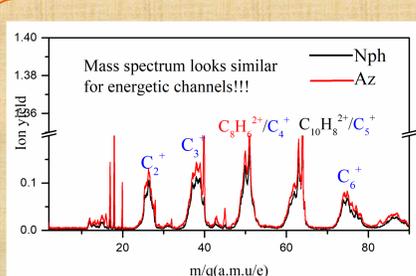
Electron emission & Electron capture cross sections of C₁₀H₈ isomers:



- Using correlated and coincidence measurements in EE, EC and CI mode detector efficiency could be obtained
- These efficiencies are then used to correct the individual peak yields.

- σ_{EE}/σ_{EC} is measured to be same for target isomers as well as for a range of proton energies
- For ion-PAH collision electron emission (EE) cross section is weakly dependent on the projectile speed and electron capture (EC) cross section shows a rapid decrease
- it is highly unlikely that two different molecules have same ratios of absolute cross sections unless their respective absolute values are equal.
- Collisional excitation in the intermediate velocity regime (1.41-2.41 a.u.) of proton is related to electron density distribution, which will be similar in molecules
- Single ionization for naphthalene is known to be dominated by plasmon resonance hence we show that it is application to azulene as well

Similarities & differences in the mass spectra of C₁₀H₈ isomers

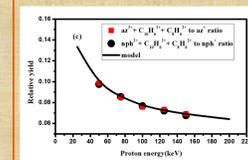
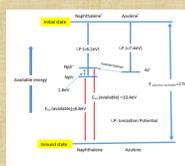


Results & discussions

- Smaller fragments produced via energetic reactions can come from equilibrium mixture of possible parent ions.
- For low energy reactions, isomerisation between cationic az and nph is a steady state reaction, which favours one direction.
- For a fixed energy excitation process (plasmon resonance at 16-17eV), $k(\text{az}^+) > k(\text{Nph}^+)$.
- As per Dykova's et al calculation, Az⁺ isomerises to Nph⁺ prior to dissociation.
- Internal energy of Nph⁺ produced from Az⁺ would have enhanced by isomerization and resulted a fast decay, it is conspicuous from the mass spectrum $E_{\text{int}}(\text{az}^+) - E_{\text{int}}(\text{nph}^+) = E_{\text{iso}} + IP_{\text{nph}} - IP_{\text{az}} = 1.6\text{eV}$

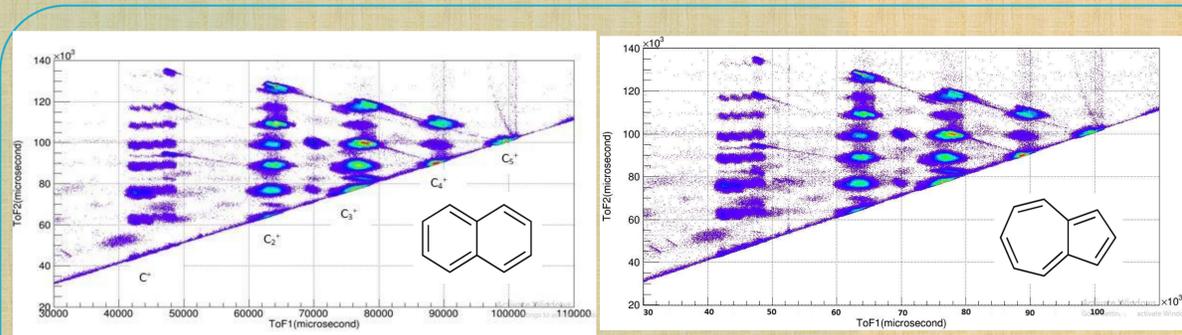
Energy diagram: plasmon resonance and isomerization

- For Nph⁺ ion formed after plasmon excitation, the internal energy can be calculated by considering the plasmon energy - adiabatic IP.
- For Nph⁺ ion formed after isomerisation of Az⁺, the internal energy can be calculated by considering the plasmon energy - adiabatic IP + isomerisation energy.

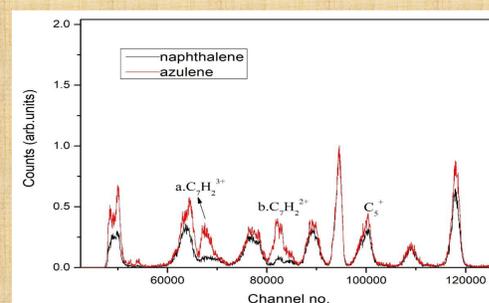


- Even the dication formation was observed to be sourced from double plasmon excitation.
- The relative cross sections of dication formation were well matched with double plasmon excitation model.

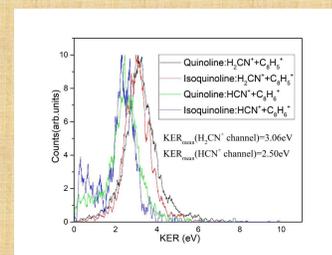
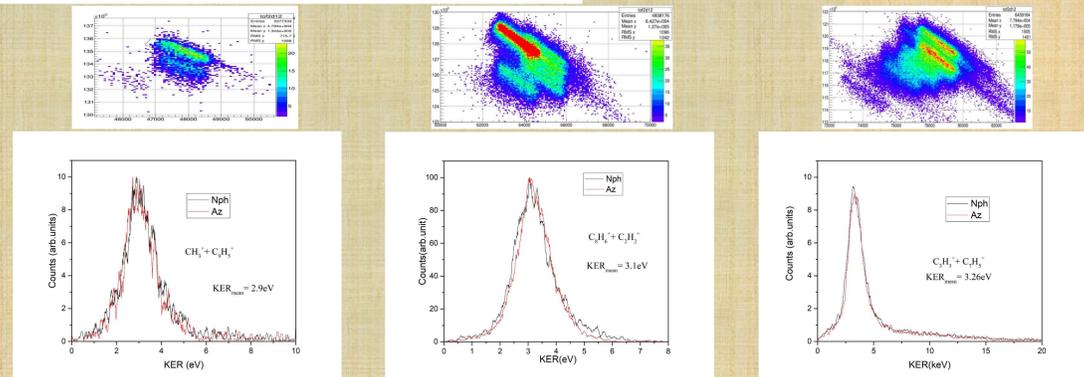
Ion-ion coincidence channels & KER values



CH₃⁺ emission channels of azulene and naphthalene



- Fragments a and b as indicated in the figure, is specific to azulene. Quadruply and triply charged products from azulene eliminate CH₃⁺ and become stable in microsecond time scale



KER spectra for PAH and PANH isomers for their binary fission channels

- A remarkable similarity is observed between ion-ion correlation spectra of isomeric PAHs & PANHs.
- Distribution of KER in the binary fission channels of isomeric PAHs and PANHs are measured to be identical.
- Identical KER spectra is obtained for quinoline and isoquinoline for the two binary channels.

Results & discussions

- KER is measured upto 3.26eV for carbon conserving channels.
- In contrast to naphthalene certain triply and quadruply charged azulene ion have higher propensity to decay via CH₃⁺ elimination.
- Emission of H₂CN⁺ from PANH dication is observed as one of the strongest channels among binary fission, corresponding molecule (amidogen) is a newly detected interstellar molecule.

- A remarkable similarity is observed between KER spectra of azulene and naphthalene.
- It indicates both molecules have same dication precursors in double product reactions.
- KER is measured upto 3.26eV for carbon conserving channels.

References

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